

**CERTIFICATE OF CHEMICAL ANALYSIS**

**EURONORM – CRM No. 577-1 FERRO-VANADIUM**

**LABORATORY MEANS (4 Values)  
 mass content in %**

| Line No              | C             | Si           | Mn            | P             | S             | Ni            | %Al<br>(Total) | Cu            | V            | Al<br>(Acid-<br>soluble) |
|----------------------|---------------|--------------|---------------|---------------|---------------|---------------|----------------|---------------|--------------|--------------------------|
| 1                    | 0.0838        | 1.726        | —             | —             | —             | 0.0458        | —              | —             | 49.98        |                          |
| 2                    | 0.0842        | 1.728        | 0.1488        | —             | 0.0298        | 0.0465        | 0.3850         | 0.0482        | 49.99        | 0.1778                   |
| 3                    | 0.0842        | 1.732        | 0.1500        | 0.0300        | 0.0320        | 0.0492        | 0.3862         | 0.0502        | 49.99        | 0.1882                   |
| 4                    | 0.0845        | 1.745        | 0.1505        | 0.0305        | 0.0320        | 0.0500        | 0.4010         | 0.0508        | 50.00        | 0.2000                   |
| 5                    | 0.0848        | 1.767        | 0.1510        | 0.0328        | 0.0322        | 0.0502        | 0.4025         | 0.0509        | 50.06        | 0.2000                   |
| 6                    | 0.0865        | 1.777        | 0.1538        | 0.0338        | 0.0327        | 0.0502        | 0.4040         | 0.0518        | 50.07        | 0.2025                   |
| 7                    | 0.0868        | 1.780        | 0.1550        | 0.0338        | 0.0332        | 0.0502        | 0.4048         | 0.0530        | 50.07        | 0.2090                   |
| 8                    | 0.0875        | 1.785        | 0.1552        | 0.0340        | 0.0332        | 0.0505        | 0.4048         | 0.0541        | 50.08        | 0.2105                   |
| 9                    | 0.0878        | 1.788        | 0.1555        | 0.0342        | 0.0333        | 0.0515        | 0.4050         | 0.0544        | 50.08        | 0.2125                   |
| 10                   | 0.0880        | 1.798        | 0.1572        | 0.0348        | 0.0336        | 0.0524        | 0.4095         | 0.0555        | 50.13        | 0.2142                   |
| 11                   | 0.0882        | 1.802        | 0.1600        | 0.0352        | 0.0339        | 0.0525        | 0.4118         | 0.0562        | 50.16        | 0.2152                   |
| 12                   | 0.0895        | 1.804        | 0.1600        | 0.0352        | 0.0339        | 0.0530        | 0.4140         | 0.0562        | 50.18        | 0.2205                   |
| 13                   | 0.0900        | 1.805        | 0.1618        | 0.0360        | 0.0346        | 0.0550        | 0.4248         | 0.0570        | 50.20        | 0.2250                   |
| 14                   | 0.0905        | 1.807        | 0.1645        | 0.0362        | 0.0350        | 0.0555        | 0.4275         | 0.0580        | 50.22        | 0.2275                   |
| 15                   | 0.0905        | 1.808        | 0.1648        | 0.0366        | 0.0352        | 0.0600        | 0.4305         | 0.0588        | 50.24        | 0.2295                   |
| 16                   | 0.0908        | 1.810        | 0.1675        | 0.0380        | 0.0360        | 0.0618        | 0.4400         | 0.0590        | 50.27        | 0.2305                   |
| 17                   | 0.0908        | 1.811        | 0.1750        | 0.0400        | 0.0360        | 0.0660        | 0.4425         | —             | 50.28        | 0.2306                   |
| 18                   | 0.0912        | 1.818        | —             | —             | 0.0360        | —             | 0.4432         | —             | 50.29        | 0.2348                   |
| 19                   | 0.0921        | 1.821        | —             | —             | 0.0365        | —             | —              | —             | 50.32        | —                        |
| 20                   | 0.0940        | 1.835        | —             | —             | 0.0367        | —             | —              | —             | 50.36        | —                        |
| 21                   | 0.0948        | 1.835        | —             | —             | 0.0378        | —             | —              | —             | 50.37        | —                        |
| <b>M<sub>M</sub></b> | <b>0.0886</b> | <b>1.790</b> | <b>0.1582</b> | <b>0.0347</b> | <b>0.0342</b> | <b>0.0530</b> | <b>0.4139</b>  | <b>0.0543</b> | <b>50.16</b> |                          |
| <b>S<sub>M</sub></b> | <b>0.0033</b> | <b>0.034</b> | <b>0.0073</b> | <b>0.0026</b> | <b>0.0020</b> | <b>0.0054</b> | <b>0.0181</b>  | <b>0.0034</b> | <b>0.13</b>  |                          |

M<sub>M</sub>: Mean of the intralaboratory means, S<sub>M</sub>: Standard deviation of the intralaboratory means

The laboratory mean values have been examined statistically to eliminate outlying values. Where a "—" appears in the table it indicates that an outlying value has been eliminated by either the Cochran or Grubbs Test. Values reported as "less than" values by the participating laboratories have not been taken into account in the statistical calculations.

**CERTIFIED VALUES**

Mass content in %

|                      | C            | Si          | Mn           | P            | S            | Ni           | %Al<br>(Total) | Cu           | V            |
|----------------------|--------------|-------------|--------------|--------------|--------------|--------------|----------------|--------------|--------------|
| <b>M<sub>M</sub></b> | <b>0.089</b> | <b>1.79</b> | <b>0.158</b> | <b>0.035</b> | <b>0.034</b> | <b>0.053</b> | <b>0.414</b>   | <b>0.054</b> | <b>50.16</b> |
| C(95%)               | 0.002        | 0.02        | 0.004        | 0.002        | 0.001        | 0.003        | 0.010          | 0.002        | 0.06         |

The half-width confidence interval C(95%) =  $\frac{t \times S_M}{\sqrt{n}}$  where "t" is the appropriate Student's t value and "n" is the number of acceptable mean values

For further information regarding the confidence interval for the certified value see ISO Guide 35:2006 sections 6.1 and 10.5.2

**This Certified Reference Material was prepared in accordance with the principles and recommendations  
 set out in ISO Guides 30 – 35 and issued by:**



**BUREAU OF ANALYSED SAMPLES LIMITED**

Newham Hall, Middlesbrough, England TS8 9EA

On behalf of:- The Iron and Steel Nomenclature Co-ordinating Committee (COCOR) of the ECISS, after approval by all the participating laboratories and all the producing organizations. (France – IRSID/CTIF, Germany – Iron and Steel CRM Working Group: Stahlinstitut VDEh, BAM Bundesanstalt für Materialforschung und –prüfung & MPI für Eisenforschung, Nordic Countries – Nordic CRM Working Group, UK – BAS Ltd.)

# EURONORM – CRM No. 577-1

## PARTICIPATING LABORATORIES

Arbed, d'Esch-Belval, Esch-sur-Alzette, Luxembourg

Böhler AG, Düsseldorf-Oberkassel, Germany

Breda Siderurgica, Milan, Italy

British Steel Corporation,

Stocksbridge and Tinsley Park Works, U.K.

Brown- Firth Research Laboratories, Sheffield, U.K

Bundesanstalt für Materialprüfung (BAM) Berlin-Dahlem, Germany

Centro Sperimentale Metallurgico (CSM), Rome, Italy

Cockerill, Seraing, Belgium

Gesellschaft für Elektrometallurgie mbH, Nürnberg, Germany

Hainaut Sambre, Couillet, Belgium

Institut de Recherches de la Sidérurgie Française (IRSID),

Saint Germain en Laye, France

Laboratoires d'Analyses Pourquery, Paris, France

London & Scandinavian Metallurgical Co. Ltd., Rotherham, U.K.

Murex Ltd., Rainham, U.K.

Ridsdale & Co. Ltd., Middlesbrough, U.K.

Société Française d'Electrometallurgie, (SOFREM) Aiguebelle, France

Société Française d'Electrometallurgie, (SOFREM) Chedde, Le Fayet, France

Société Nouvelle des Acieries de Pompey, France

Sollac, Florange, France

Stahlwerke Röchling-Burbach GmbH, Völklingen-Saar, Germany

Thyssen Edeltahlwerke AG, Forschungsinstitut, Krefeld, Germany

## METHODS USED

| Element                           | Line Number                               | Methods   |  |
|-----------------------------------|---|---|--|
| <b>C</b>                          | 1-3-4-15                                  | Combustion, thermal conductivity  |  |
|                                   | 2-10-13-17                                | Combustion, gas conductivity  |  |
|                                   | 5-9                                       | Combustion, coulometric titration   |  |
|                                   | 6-7-8-12-14-19-20-21                      | Combustion, infrared absorption   |  |
|                                   | 11-16                                     | Combustion, non-aqueous titration   |  |
|                                   | 18  | Combustion, gravimetric   |  |
|                                   | <b>Si</b>                                 | 1-3-6-9-11-15-17-19-21  | Gravimetry, dehydration with perchloric acid                                   |
|                                   |   | 2-4-5-7-8-10-12-13-14-16-20   | Gravimetry, dehydration with sulphuric acid                                    |
| 18                                |   | Acidimetric titration of fluosilicate   |  |
| <b>Mn</b>                         |   | 2-4-5-7-9-10-11-12-13-14-15-16-17   | Atomic absorption spectrometry   |
|                                   | 3-8                                       | Spectrophotometry, periodate oxidation  |  |
|                                   | 6   | Spectrophotometry, persulphate oxidation  |  |
|                                   | <b>P</b>                                  | 3-4-6-9   | Spectrophotometry, molybdenum blue, extraction                                 |
|                                   |   | 5-8-11-13-14-17   | Spectrophotometry, phosphovanadomolybdate, extraction                          |
|                                   |   | 7-10  | Spectrophotometry, molybdenum blue, without extraction                         |
| 12-15                             |   | Spectrophotometry, phosphovanadomolybdate, without extraction                           |  |
| 16                                | Acidimetric titration as phosphomolybdate |   |  |
| <b>S</b>                          | 2-10                                      | Combustion, conductivity  |  |
|                                   | 3-4-5-6-9-11-12-13-15-16                  | Combustion, infrared absorption   |  |
|                                   | 7-17-18                                   | Combustion, acidimetric titration   |  |
|                                   | 8   | Combustion, coulometric titration   |  |
|                                   | 14  | Combustion, spectrophotometry, para rosaniline  |  |
|                                   | 19-20-21                                  | Combustion, oxidation reduction titration   |  |
|                                   | <b>Ni</b>                                 | 1   | Titration with cyanide, separation with dimethylglyoxime                       |
| 2-3-7-8-9-10-11-12-13-14-15-16-17 |   | Atomic absorption spectrometry  |  |
| 4-5                               |   | Spectrophotometry, dimethylglyoxime   |  |
| 6                                 |   | Spectrophotometry, dimethylglyoxime after ion exchange separation                       |  |
| <b>Al<br/>(Total)</b>             |   | 2   | Spectrophotometry, hydroxyquinolate, fusion of insoluble residue with peroxide |
|                                   | 3   | Atomic absorption spectrometry  |  |
|                                   | 4-9-14-15                                 | Atomic absorption spectrometry, fusion of insoluble residue with bisulphate             |  |
|                                   | 5   | Spectrophotometry, eriochrome cyanine, fusion of insoluble residue with carbonate/borax |  |
|                                   | 6-16                                      | Atomic absorption spectrometry, fusion of insoluble residue with pyrosulphate           |  |
|                                   | 7   | Spectrophotometry, chrome azurol S, fusion of insoluble residue with peroxide           |  |
|                                   | 8-10-11-12-17-18                          | Atomic absorption spectrometry, fusion of insoluble residue with carbonate/borax        |  |
|                                   | 13  | Atomic absorption spectrometry, fusion of insoluble residue with peroxide               |  |
|                                   | <b>Cu</b>                                 | 2-3-4-5-7-8-10-11-12-14-15-16   | Atomic absorption spectrometry   |
|                                   |   | 6-9   | Spectrophotometry, diethyldithiocarbamate                                      |
| 13                                |   | Spectrophotometry, 2,2 diquinolyl   |  |
| <b>V</b>                          | 1-2-3-5-6-8-15-16-21                      | Titration with Fe(II), visual end point   |  |
|                                   | 4-7-9-10-11-12-14-17-18-19-20             | Titration with Fe(II), potentiometric end point   |  |
|                                   | 13  | Titration with Mn (VII), potentiometric end point                                       |  |
|                                   | <b>Al<br/>(Acid Soluble)</b>              | 2-3-4-5-6-7-10-11-12-14-15-16-17  | Atomic absorption spectrometry   |
| 8                                 |   | Spectrophotometry, chrome azurol S  |  |
| 9                                 |   | Spectrophotometry, eriochrome cyanine after mercury cathode separation                  |  |
| 13                                |   | Gravimetry as oxinate   |  |
| 18                                |   | Complexometric titration after ion exchange separation                                  |  |

## DESCRIPTION OF THE SAMPLE

The sample consists of a fine powder which has been sieved to a nominal particle size of 50 to 180 microns. It is supplied in bottles containing 100g.

## INTENDED USE & STABILITY

ECRM 577-1 is intended for the verification of analytical methods, such as those used by the participating laboratories, for the calibration of analytical instruments in cases where calibration with primary substances (pure metals or stoichiometric compounds) is not possible and for establishing values for secondary reference materials.

It will remain stable provided that the bottle remains sealed and is stored in a cool, dry atmosphere. When the bottle has been opened the lid should be secured immediately after use. If the contents should become discoloured (e.g. oxidised) by atmospheric contamination they should be discarded.

## TRACEABILITY

The traceability of ECRM 577-1 has been established in accordance with principles of ISO Guides 30 – 35 and the International Vocabulary of Basic and General Terms In Metrology.

The characterisation of this material has been achieved by inter-laboratory study, each laboratory using the method of their choice, details of which are given above. These methods are either stoichiometric analytical techniques or methods which are calibrated against pure metals or stoichiometric compounds. Most methods used were either international or national standard methods or methods which are technically equivalent.

## FURTHER INFORMATION

For information regarding the preparation, certification and supply of these European Certified Reference Materials (EURONORM-CRMs) and the use of the statistical information given on this certificate, please refer either to the producer of this Certified Reference Material or to Technical Reports CEN/TR 10317:2013 and CEN/TR 10350:2013, both of which are available from the national standards body in your country. (In the UK this is the BSI, 389 Chiswick High Road, London W4 4AL).

Further information and advice on this or other Certified Reference Materials or Reference Materials produced by Bureau of Analysed Samples Ltd. may be obtained from the address below.

Pour disposer d'informations sur la fabrication, la certification et la distribution des Matériaux de Référence Certifiés Européens (EURONORM-MRC) ainsi que sur l'utilisation des informations statistiques données sur ce certificat, se reporter soit au producteur de ce Matériau de Référence Certifié, soit aux Rapports Techniques CEN/TR 10317:2013 et CEN/TR 10350:2013. On peut se procurer ces deux documents auprès des organismes nationaux de normalisation. (Pour la France: AFNOR, 11 Avenue Francis de Pressensé, 93571 – St Denis la Plaine Cedex).

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NEWHAM HALL, NEWBY,  
MIDDLESBROUGH, ENGLAND, TS8 9EA  
Email: enquiries@basid.co.uk  
Website: www.basid.co.uk

For BUREAU OF ANALYSED SAMPLES LTD.

R P MEERES,  
MANAGING DIRECTOR